

<u>MENG 3401 – Thermodynamics</u> <u>Course Syllabus</u>

Com orton /	E-11 2022
Semester /	Fall 2023
Year	
Catalog	Thermodynamic properties of pure substances. Definitions of work, heat, and energy.
Description	First and second laws of thermodynamics and its application to fixed mass systems and
D	control volumes. Analysis of thermodynamic cycles and their components.
Prerequisites	C or better grade in ENGR 2302 Dynamics, PHYS 2325 Physics I, PHYS 2125 Physics
G 4'	I Lab
Section Number	030
	Hander Abdul Derreh DhD DE
Instructor Contact info	Hayder Abdul-Razzak, PhD, PE
	habdulrazzak@uttyler.edu
Class Type /	Face to Face, A218
Location	
Class Times	Mon/Wed 5:00PM to 6:50PM
Office Hours	Mon/Wed 2:00PM to 3:30PM or by appointment
Credits	4
Textbooks	1. Fundamentals of Engineering Thermodynamics, 8th ed., by Moran, Shapiro, et
and	al., John Wiley and Sons, zyBook ISBN: 979-8-203-18310-1)
Reference	2. Instruction to subscribe to Thermodynamics zybook will be provided in Canvas
Materials	
Optional References	N/A
Additional	N/A
	N/A
requirements Instruction /	- 4 Quizzes (50 x 4) 200 points
Evaluation /	
Evaluation Method/	- 4 Exams (4 x 100) 400 points
Homework	Practice questions shall be assigned but not graded. Students must turn in the homework
HUIICWULK	during the class period on the due date; please keep in mind that no late submission will
	be accepted.
	The revised homework will be needed to discuss problems of exams if the student asks
	for review. That is, if a problem in the exam has a related problem in a homework, the
	student waives the right of discussing the problem with the instructor if the student does
	not present/bring the revised homework problem.
Grading	Grading in this course will be based on the following:
Policy / Scale	Scale: $A = > 90, B = > 80, C = > 70, D = > 60, F < 60.$
	Grade appeal: grades can be appealed by meeting the instructor during office hours, but
	no later than a week after the grade has been given.
Important	See UT Tyler Academic Calendar: <u>https://www.uttyler.edu/schedule/files/2023-</u>
events/dates	2024/academic-calendar-2023-2024-main-20230614b.pdf



Attendance / Makeup policy	Attendance at every meeting is strongly encouraged but not mandatory. There will be no makeup for missed in-class work. An opportunity to make up a missed exam may be available to students with an excused absence. Be advised that makeup exams maybe more challenging. Excused absences include absences for University- sponsored events and for religious observances (see the University policy link above for the procedures to follow). Other makeups are granted only in extreme cases and at the discretion of the instructor. Excused absence due to illness will require evidence of treatment by medical personnel or at a medical facility.
Course	personnel or at a medical facility.
Course Learning	By the end of this course students will be able to: 1. Determine properties of substances (Applying appropriate physical models of
Objectives /	state for a substance).
ABET &	 Calculate the work done by and heat taken in by a system undergoing a change
PEOs	of state (reversibly and irreversibly).
relation	3. Perform first and second law analysis of steady-state flow systems (heat
	exchangers, turbines, pumps, condensers, boilers, and throttle valves).
	4. Perform analysis of thermodynamic cycles (e.g. Carnot, Rankine and Brayton
	cycles).
	5. Perform psychrometric analysis for heating/cooling processes.
Course	• Equations of state and physical principles behind liquid/gas phase separation.
Outline	Relationship between pressure/volume, temperature/volume, and
	pressure/temperature spaces.
	• Computation of mechanical work and relation to pressure/volume space.
	• Designation of global/macroscopic kinetic and potential energy and internal
	energy as a property of state.
	• First law and computation of heat transfer.
	• Measurement of heat transfer and conversion to an "equivalent" work.
	• First law analysis of steady state flow systems: turbines, pumps/compressors, throttles, boilers, nozzles, diffusers, single substance mixing chambers, and heat exchangers.
	• Irreversibility and definition of entropy.
	• Quantification of entropy.
	• Forms of the second law: entropy statement and logical equivalence with
	Clausius and Kelvin-Planck statements.
	• Definition of cycle efficiency and comparison with theoretical limit (Carnot).
	• Second law analysis of steady state flow systems: turbines, pumps/compressors,
	throttles, boilers, nozzles, diffusers, single substance mixing chambers, and heat
	exchangers.
	• Isentropic efficiency of turbines and pumps/compressors.
	• Efficiency of Rankine and Brayton cycles.
	• Vapor phase cycle/Refrigeration cycle and Heat Pump Systems.
	Psychrometry
University	https://www.uttyler.edu/academic-affairs/files/syllabus_information_2021.pdf
Policies	